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Информация о владельце:

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Дата подписания: 19.05.2023 16:30:35 PEOPLES' FRIENDSHIP UNIVERSITY OF RUSSIA **RUDN** University

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educational division (faculty/institute/academy) as higher education programme developer

COURSE SYLLABUS

Organic chemistry course title Recommended by the Didactic Council for the Education Field of: 36.05.01 Veterinary field of studies / speciality code and title

The course instruction is implemented within the professional education programme of higher education:

36.05.01 Veterinary

higher education programme profile/specialisation title

1. GOALS AND OBJECTIVES OF THE COURSE

The aim of mastering the course "Organic chemistry" is to familiarize students with the theoretical foundations of organic chemistry, the most important practical applications, without which it is impossible to solve modern technological, environmental problems, understanding of the processes occurring in living organisms.

2. REQUIREMENTS FOR LEARNING OUTCOMES

The implementation of the course "Organic chemistry" is aimed at creating the following competencies (parts of competencies) for students:

Table 2.1. List of competencies formed by students during the development of the

Competence	Competence descriptor	Indicators of competence		
code		accomplishment (within the course)		
GC-8	Is able to create and maintain safe living conditions in everyday life and professional activities to preserve the natural environment, ensure the sustainable development of society, including the threat and emergence of emergencies and military conflicts	GC-8.1 Analyzes factors of harmful influence on the life activity of elements of the environment (technical means, technological processes, materials, buildings and constructions, natural and social phenomena) GC-8.2 Identifies hazardous and harmful factors within the scope of the job		
GPC-4	Is able to use in professional activity methods to solve problems using modern equipment in the development of new technologies and use modern professional methodology to conduct experimental research and interpretation of the results	GPC-4.1 Has the conceptual and methodological apparatus of the basic natural sciences at a level sufficient for full professional activity at the modern level GPC-4.3 Willing to use modern methodology in designing and conducting experimental research GPC-4.4 Uses modern professional methodology in interpreting research results		

3. COURSE IN HIGHER EDUCATION PROGRAMME STRUCTURE

The course "Organic chemistry" refers to the mandatory part of block B1 of the Educational Program of Higher Education.

As part of the Educational Program of Higher Education, students also master other courses and /or practices that contribute to achieving the planned results of mastering the course "Organic chemistry".

Table 3.1. List of Higher Education Program components courses that contribute to expected learning outcomes

Competence code	Competence descriptor	Previous courses/modules, internships*	Subsequent courses/modules, internships*
GC-8	Is able to create and maintain safe living conditions in everyday life and professional activities to preserve the natural environment, ensure the sustainable development of society, including the threat and emergence of emergencies and military conflicts	Basics of Professional Ethics Inorganic and analytical chemistry	Biological physics Life safety Veterinary Microbiology and Mycology Virology and biotechnology Veterinary radiobiology General and Veterinary Ecology Study practice Preparation for and passing the state exam
GPC-4	Is able to use in professional activity methods to solve problems using modern equipment in the development of new technologies and use modern professional methodology to conduct experimental research and interpretation of the results	Inorganic and analytical chemistry	Biological physics Physical and Colloidal Chemistry Biological chemistry Maths Immunology Laboratory diagnostics of infectious and invasive diseases Veterinary and industrial laboratories with design basics Study practice Clinical internship Industrial practice Academic research practice with the preparation of a scientific qualification project Preparation for and passing the state exam

4. COURSE WORKLOAD AND TRAINING ACTIVITIES

Course workload of the course "**Organic chemistry**" is 2 credits.

Table 4.1. Types of academic activities during the period of the HE program mastering for **full-time** study

Types of academic activities	HOURS	Semesters			
Types of academic activities		2	-	-	ı
Contact academic hours	36	36	-	-	-

including						
Lectures		18	18	-	-	-
Lab work	18	18	-	-	-	
Seminars (workshops/tutorials)		-	-	-	-	-
Self-study		26	26	-	-	-
Evaluation and assessment (exa	10	10	-	-	-	
grading)						
	Academic	72	72	-	-	-
Course workload						
Course workload	Credit	2	2	_	-	_
	unit					

5. COURSE CONTENTS

Table 5.1 Content of the course (module) by type of academic work

Modules Content of the modules (topics) Types of		
Modules	Content of the modules (topics)	Types of
		academic
		activities
Module 1. Introduction	Topic 1.1. The subject of organic	
	chemistry. Carbon compounds, their	work.
	characteristics, natural sources of organic	
	compounds. The importance of organic	
	chemistry as a tool of knowledge of man's	
	technogenic influence on the	
	environment. Brief sketch of the history	
	of organic chemistry.	
	The theory of structure of organic	
	compounds (Butlerov A.M.), the present	
	state of the theory of chemical structure.	
	Principles of nomenclature of organic	
	compounds. Nomenclature of UPAC.	
	Classification of organic compounds.	
	Rows, classes, functional groups.	
	Basic principles of qualitative and	
	quantitative analysis, methods of	
	establishing the structure of organic	
	compounds.	
Module 2. Hydrocarbons.	Topic 2.1. Alkanes. Homological series.	Lectures, Lab
	Nomenclature, isomerism, methods of	
	preparation of alkanes. Physical	
	properties. Chemical properties.	
	Identification of alkanes.	
	Topic 2.2. Alkenes. Homological series,	
	nomenclature. Isomerism. Methods for	
	obtaining alkenes. Physical properties.	
	Chemical properties: electrophilic	
	mechanism of addition to alkenes.	
	Markovnikov's rule. Radical addition in	
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		the presence of peroxides (Harash).		
		Identification of alkenes.		
		Topic 2.3 Alkynes. Homological series,		
		nomenclature. Methods for preparation of		
		alkynes. Physical properties. Chemical		
		properties. Adhesion reactions.		
		Dimerization of acetylene. Reactions of		
		acetylene hydrogen atom: formation of		
		acetylenides. Identification of alkynes.		
		Topic 2.4. Diene hydrocarbons.	Lectures,	Lab
		Homological series, classification and		
		nomenclature. Electronic structure of		
		conjugated double bond system. Methods		
		of preparation of divinyl, isoprene and		
		chloroprene. Chemical properties of		
		conjugated dienes: reactions of addition		
		to 1,2- and 1,4- positions; polymerization		
		reactions. Rubber (NK, SK) and plastics.		
		Identification of dienes.		
Module 3:	Aromatic	Module 3.1. Aromatic hydrocarbons	Lectures,	Lab
hydrocarbons	and	(arenes). Homological series,	work.	
homofunctional		nomenclature and isomerization of		
compounds.		benzene hydrocarbons. Electronic		
		structure of the benzene molecule.		
		Aromaticity, Hückel rule. Methods for		
		obtaining arenes, their physical		
		properties. Chemical properties:		
		electrophilic substitution of hydrogen in		
		the benzene nucleus. Mechanism of		
		reaction. Orientation rules for		
		electrophilic substitution: ortho- and		
		meta-orientants and their influence on		
		subsequent substitution in the benzene		
		core. Condensed aromatic systems.		
		Methods for the identification of arenes.		
		Module 3.2. Halogen derivatives.	Lectures,	Lab
		Nucleophilic substitution reactions of	work.	
		halogen in halide alkyls and arynes. SN1		
		and SN2 - Mechanisms of substitution.		
		Elimination reactions. Zaitsev's rule.		
		Organometallic compounds. Comparison		
		of chemical activity of halogen bound to		
		carbon of benzene ring with carbon of		
		side cycle. Identification of halogen		
		derivatives of HC.		
		Module 3.3. Alcohols. Classification,	· · · · · · · · · · · · · · · · · · ·	Lab
		nomenclature and isomerism. Methods		
		for the production of alcohols. Physical		

	properties, hydrogen bonds. Chemical		
	properties of monatomic alcohols. Simple		
	esters. Preparation, properties and		
	applications. Bi-atomic alcohols		
	(glycols). Preparation, chemical		
	properties, applications.		
	Three-atom alcohols (glycerols). Natural		
	sources and chemical methods of		
	production. Properties and applications of		
	glycerol. Phenols. Nomenclature and		
	isomerization. Methods of production.		
	Physical properties. Electronic structure		
	of phenol molecule. Influence of		
	substituents in benzene ring on acid		
	properties of phenols. Chemical		
	properties of phenols. Electrophilic		
	substitution reactions in the benzene ring		
	of phenols. Phenol-formaldehyde resins.		
	Identification of alcohols and phenols.	-	
	Module 3.4. Amines. Classification,		Lab
	nomenclature, isomerism. Methods for	work.	
	preparation of amines. Physical		
	properties. Chemical properties salt		
	formation, alkylation, acylation, action of		
	nitric acid on amines.		
	Aromatic amines. Aniline, methods of its		
	preparation. Substitution reactions of		
	aromatic amines in the nucleus and		
	reactions by amino group. Comparison of		
	basic properties of fatty and aromatic		
	amines. Identification of amines.		
	Module 3.5. Aldehydes and ketones.	Lectures	Lab
	Isomerism and nomenclature. Methods of		Lau
		WOIK.	
	production. Structure of the carbonyl		
	group. Physical properties. Chemical		
	properties: reactions of nucleophilic		
	addition to carbonyl group. Substitution		
	reactions of carbonyl oxygen. Haloform		
	reaction. Reaction of formation of acetals		
	(catalysts). Reactions involving hydrogen		
	in the α -position to the carbonyl group.		
	Aldole and croton condensations.		
	Reduction and oxidation of aldehydes		
	and ketones. Identification of oxo		
	compounds.		
Module 4. Carboxylic	Module 4.1. Carboxylic acids. Isomerism	Lectures,	Lab
Acids and Heterofunctional	I	work.	
Compounds	carboxylic group. Influence of the		
Composition	The Stoap. Initiative of the	<u> </u>	

	structure of carboxylic acids on their acidic properties. Methods for production. Physical properties. Chemical properties: reactions by carboxylic group and by α-position to carboxylic group. Derivatives of carboxylic acids: halogenanhydrides, anhydrides, nitriles, amides, esters.		
	Module 4.2. Lipids. Natural fats and oils - glycerides of higher fatty acids. Hydrolysis of fats, soaps. Hydrogenation of fats, margarine.	Lectures, L work.	∠ab
	Module 4.3. Non-saturated carboxylic	Lectures, L work.	ab
	Module 4.4. Bivalent carboxylic acids, methods of their production, properties and applications. Unsaturated bivalent acids.	· ·	Lab
	Module 4.5. Oxic acids. Basicity and atomicity. Methods of preparation. General and specific properties of oxyacids. Salicylic acid. Relation of α -, β - and γ -oxy acids to heating.		ab
	Module 4.6. Oxo acids (aldehyde and keto acids). Nomenclature, structure and methods of production. Chemical properties.		∟ab
	Module 4.7. Amino acids. Classification, nomenclature, structure and methods of production of amino acids. Isoelectric current. Chemical properties of amino acids, transformations by heating of α -, β - and γ -amino acids. Peptides.		ab
Module 5. Carbohydrates	Module 5.1. Monosaccharides: aldoses and ketoses, isomerism, configuration. Ring-chain tautomerism of monoses. Mutarotation. Reactions of monoses by carbonyl and oxy groups.	•	∟ab

Classroom for Academic Activity Type	Equipping the classroom	Specialized educational/laboratory equipment, software and materials for the development of the course (if necessary)
Lecture	An auditorium for conducting lecture-type classes, equipped with a set of specialized furniture; a board (screen) and technical means of multimedia presentations.	BENQ MX661 projector, NEC NP40 projector, motorized screen for projectors
Laboratory	An auditorium for laboratory work, individual consultations, routine monitoring and interim certification, equipped with a set of specialized furniture and equipment.	specialized equipment of the chemical laboratory: fume hood SHVP-4 (6 pcs.), rotary evaporator Hei-value digital G3B, rotary evaporator IKA, digital instruments to determine the melting point SMP10, electronic laboratory scales AND EK-610, MK-M flask heaters of different volumes, drying oven PE-4610, magnetic stirrer MRHei-Mix S, magnetic stirrer with heating MRHei-Standart, Refractometer, combined laboratory bath, chemical vacuum station PC3001 VARIO-pro. RZ2.5 rotary vane vacuum pump, MZ2CNT chemistry diaphragm vacuum pump, Steinel air blower, Spectroline EB-280C UV lamp, chemical glassware, refrigerator
Self-studies	An auditorium for independent work of students (can be used for seminars and consultations), equipped with a set of specialized furniture and computers with access to an electronic information and educational environment.	-

7. RESOURCES RECOMMENDED FOR COURSE STUDIES

Main readings:

1. Grandberg Igor Johannovich. Organic Chemistry: textbook / I.I. Grandberg, N.L.

- Nam. 10th ed. Saint-Petersburg: Lan' Publisher, 2021. 608 p.: ill. (Higher Education). 978-5-8114-8835-3: 2571.25.00.
- 2. Fundamentals of organic chemistry: textbook for students in pharmacy / T.N. Borisova, A.V. Varlamov, E.A. Sorokina [etc.]. 2nd ed. amended; Electronic text data. M.: RUDN, 2019. 355 p.: ill. ISBN 978-5-209-09033-5: 352.76.
- organic chemistry: educational and methodical manual for laboratory works for students of the 1st year, studying on the specialty "Ecology and nature management" / E. V. Nikitina, E. A. Sorokina, F. I. Zubkov, L. N. Kulikova. Electronic text data. M.: RUDN, 2019. 36 c. ISBN 978-5-209-09035-9: http://lib.rudn.ru/MegaPro/Web

Additional Readings:

- 1. Questions and Problems in Organic Chemistry: Textbook / Compiled by T.N. Borisova, A.A. Varlamov, E.A. Sorokitina, E.A. Nikitina. T. N. Borisova, A. V. Varlamov, E. A. Sorokina, E. V. Nikitina. 3rd ed. M.: RUSSIAN ASSOCIATION OF RUSSIAN TRADE UNIONS, 2020. 97 c. 978-5-209-09582-8: 79.40.
- organic chemistry. Tasks for the general course with solutions: a textbook in 2 parts. Part 1 / M.V. Livantsov, G.S. Zaitseva, L.I. Livantsova [et al]; ed. by N.S. Zefirov. Ed. 3-th edition; Electronic text data. M.: Laboratory of knowledge, 2019. 255 p.: ill. (Textbook for higher school). ISBN 978-5-00101-174-3: http://lib.rudn.ru/MegaPro/Web

Internet sources

- 1. Electronic libraries (EL) of RUDN University and other institutions, to which university students have access on the basis of concluded agreements:
- RUDN Electronic Library System (RUDN ELS) http://lib.rudn.ru/MegaPro/Web
- EL "University Library Online" http://www.biblioclub.ru
- EL "Yurayt" http://www.biblio-online.ru
- EL "Student Consultant" www.studentlibrary.ru
- EL "Lan" http://e.lanbook.com/
- EL "Trinity Bridge"

2. Databases and search engines:

- electronic foundation of legal and normative-technical documentation http://docs.cntd.ru/
- Yandex search engine https://www.yandex.ru/
- Google search engine https://www.google.ru/
- Scopus abstract database http://www.elsevierscience.ru/products/scopus/

Educational and methodological materials for independent work of students during the development of the course/ module*:

- 1. A course of lectures on the course "**Organic chemistry**".
- 2. Laboratory workshop on the course "Organic chemistry".
- * The training toolkit and guidelines for the internship are placed on the internship page in the university telecommunication training and information system under the set procedure.

8. ASSESSMENT TOOLKIT AND GRADING SYSTEM* FOR EVALUATION OF STUDENTS' COMPETENCES LEVEL AS COURSE RESULTS

The assessment toolkit and the grading system* to evaluate the level of competences (competences in part) formation as the course results are specified in the Appendix to the course syllabus.

* The assessment toolkit and the grading system are formed on the basis of the requirements of the relevant local normative act of RUDN University (regulations / order).

DEVELOPER:		
Associate Professor in the Department of Organic		
Chemistry		Kulikova L.N.
Position, Basic curriculum	Signature	Full name.
HEAD OF EDUCATIONAL DEPARTMENT:		
Department of Organic Chemistry		Voskresensky L.G.
Name Basic Curriculum	Signature	Full name.
HEAD OF		
HIGHER EDUCATION PROGRAMME:		
Director of the Department of Veterinary Medicine		Vatnikov Yu.A.
Position Basic curriculum	Signature	Full name