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(RUDN University) Science faculty

educational division (faculty/institute/academy) as higher education programme developer

COURSE SYLLABUS

THEORETICAL ORGANIC CHEMISTRY

course title

Recommended by the Didactic Council for the Education Field of:

04.04.01 «Chemistry»

field of studies / speciality code and title

The course instruction is implemented within the professional education programme of higher education:

«Fundamental and applied chemistry»

higher education programme profile/specialisation title

1. COURSE GOAL(s)

The goal of the course of «Theoretical organic chemistry» is in the formation of the system of knowledge about the laws in the chemical behavior of the main classes of organic compounds in relation to their structure in order to use this knowledge as a basis to solve specific problems.

2. REQUIREMENTS FOR LEARNING OUTCOMES

Mastering the discipline "Theoretical organic chemistry" expects students to acquire the following competences /(competences in part):

Table 2.1. List of competences that students acquire through the course study

Competence code	Competence descriptor	Competence formation indicators (within this course)
M-PC-2-s	Ability to conduct patent information research in the chosen field of chemistry and/or related sciences	M-PC-2-s-2 Analyzes and summarizes the results of a

3. COURSE IN HIGHER EDUCATION PROGRAMME STRUCTURE

The course refers to the elective component of (B1) block of the higher educational programme curriculum.

Within the higher education programme students also master other (modules) and / or internships that contribute to the achievement of the expected learning outcomes as results of the course study.

Table 3.1. The list of the higher education programme components/disciplines that contribute to the achievement of the expected learning outcomes as the course study results

Competenc	Competence	Previous	Subsequent	
e code	descriptor	courses/modules*	courses/modules*	
-	-		courses/modules* Domino reactions in the synthesis of heterocycles NMR of organic compounds Molecular spectral analysis Chemistry of natural compounds Chemistry of heterocyclic compounds Mass spectrometry of organic compounds Stereochemistry Drug design basics	
			Experimental research methods in chemistry	
			Scientific research practice Undergraduate practice	

* To be filled in according to the competence matrix of the higher education programme.

4. COURSE WORKLOAD AND ACADEMIC ACTIVITIES

The total workload of the course is 3 credits.

Table 4.1. Types of academic activities during the periods of higher education programme mastering (full-time training)

Type of academic activities		Total	Semester(-s)			
		academic hours	1	2	3	4
Contact academic hours		36				
including:						
Lectures (LC)		36	36			
Lab Works (LW)						
Seminars (workshops/tutorials) (S)						
Self-studies		54	54			
<i>Evaluation and assessment (exam/passing/failing grade)</i>		18	18			
Course workload	academic hours	108	108			
	credits	3	3			

5. COURSE CONTENTS

Course module title	Course module contents (topics)	Academic activities types*
Section 1. Chemical bond	Theme 1.1. Types of chemical bonds. Hybridization. Localized and delocalized chemical bonds. Multicenter bonds. Method of molecular orbitals.	LC
in organic compounds. Electronic effects.	Theme 1.2. Inductive effect and conjugation effect. The effects of superconjugation. Dependence of effects on the structure of molecules.	LC
Section 2. Acid-base	soft acids and bases	
properties and spatial structure of organic compounds.	Theme 2.2. Conformations of acyclic and cyclic molecules. Influence of conformation on reactivity. Chirality and symmetry. Optical activity. Types of chiral molecules. Enantiomers and diastereomers.	LC
Section 3. Mechanisms of organic reactions. Nucleophilic substitution in the aliphatic series, Elimination and addition	Theme 3.1. General ideas about the mechanisms of organic reactions. Intermediate particles in transformations of organic compounds. Methods for establishing and studying the mechanisms of organic reactions.	LC
reactions by multiple bonds.	Theme 3.2. Reactions $S_N 1$, $S_N 2$, $S_N i$. Influence of the structure, substrate and reaction conditions on the mechanism. E1 and E1cB mechanisms, E2-mechanism. Factors affecting the mechanism of	LC

Table 5.1. Course contents and academic activities types

Course module title	Course module contents (topics)	Academic activities types*	
	cleavage reactions. Mechanisms of electrophilic addition at C=C-bond and nucleophilic at C=O- bond. The role of the acidity of the medium upon addition to C=O.		
Section 4. Aromaticity. Substitution in the	Theme 4.1. Types of aromatic systems. Aromaticity criteria. Antiaromatic. Electrophilic substitution: reagents, π - and σ -complexes. Nucleophilic substitution: mechanism of the process, Meisenheimer complexes. Arin mechanism.	LC	
aromatic series. Pericyclic reactions. Rearrangements	Theme 4.2. [4+2]-Cycloaddition, process synchronism, influence of substituents. Woodward-Hoffmann rules. Rearrangements: nucleophilic, electrophilic and free-radical. Cope and Claisen rearrangements.	LC	

* - to be filled in only for <u>full</u>-time training: *LC* - *lectures; LW* - *lab work; S* - *seminars.*

6. CLASSROOM EQUIPMENT AND TECHNOLOGY SUPPORT REQUIREMENTS

Type of academic activities	Classroom equipment	Specialised educational / laboratory equipment, software, and materials for course study (if necessary)
Lecture	A lecture hall for lecture-type classes, equipped with a set of specialised furniture; board (screen) and technical means of multimedia presentations.	
Self-studies	A classroom for independent work of students (can be used for seminars and consultations), equipped with a set of specialised furniture and computers with access to the electronic information and educational environment.	

Table 6.1. Classroom equipment and technology support requirements

* The premises for students' self-studies are subject to **MANDATORY** mention

7. RESOURCES RECOMMENDED FOR COURSE STUDY

Main readings:

1. M. B. Smith, J. March March's Advanced Organic Chemistry: Reactions, Mechanisms, and Structure, 6th Edition, Wiley-Interscience, 2007.

Additional readings:

1. J. Hine Physical Orgamic Chemistry, McGraw-Hill, 2013.

2. D.A. Dougherty, E.V. Anslyn Modern Physical Organic Chemistry, Univ. Sci. Books, 2006.

3. J. Clayden, N. Greeves, S. Warren, P. Wothers Organic Chemistry, Oxford Univ. Press. 2001.

4. T.H. Lowry, K.S. Richardson Mechanism and Theory in Organic Reactions. 3rd Ed. Benjamin Cumming Publ., 1987.

Internet sources

1. Electronic libraries (EL) of RUDN University and other institutions, to which university students have access on the basis of concluded agreements:

- RUDN Electronic Library System (RUDN ELS) http://lib.rudn.ru/MegaPro/Web

- EL "University Library Online" <u>http://www.biblioclub.ru</u>
- EL "Yurayt" http://www.biblio-online.ru
- EL "Student Consultant" www.studentlibrary.ru
- EL "Lan" http://e.lanbook.com/
- EL "Trinity Bridge"

2. Databases and search engines:

- Yandex search engine https://www.yandex.ru/
- Google search engine <u>https://www.google.ru/</u>
- abstract database SCOPUS http://www.elsevierscience.ru/products/scopus/
- database Reaxys https://www.reaxys.com/#/search

Training toolkit for self- studies to master the course *:

1. The set of lectures on the course "Theoretical organic chemistry".

* The training toolkit for self- studies to master the course is placed on the course page in the university telecommunication training and information system under the set procedure.

8. ASSESSMENT TOOLKIT AND GRADING SYSTEM* FOR EVALUATION OF STUDENTS' COMPETENCES LEVEL UPON COURSE COMPLETION

The assessment toolkit and the grading system* to evaluate the competences formation level (competences in part) upon the course study completion are specified in the Appendix to the course syllabus.

* The assessment toolkit and the grading system are formed on the basis of the requirements of the relevant local normative act of RUDN University (regulations / order).

DEVELOPERS:

Associate Professor of Organic Chemistry Department	any	N. E. Golantsov		
Position, Department	Signature	name and surname		
HEAD OF EDUCATIONAL DEPA	ARTMENT:			
Organic Chemistry Department	Doander	L. G. Voskressensky		
Name of Department	Signature	name and surname		
HEAD				
OF HIGHER EDUCATION PROC	GRAMME:			
Dean of Science faculty,				
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