Документ подписан простой электронной подписью Информация о владельце:

ФИО: Ястребов Олег Арексиндрам State Autonomous Educational Institution of Higher Education Должность: Ректор PEOPLES' ERIENDSHIP UNIVERSITY OF RUSSIA named after P. Lumumba
Дата подписания. 27.05.2023 17.58.22 (RUDN University)

Уникальный программный ключ:

ca953a0120d891083f939673078ef1a989dae18a

Science faculty

educational division (faculty/institute/academy) as higher education programme developer

COURSE SYLLABUS
Chemistry of natural compounds
course title
Recommended by the Didactic Council for the Education Field of:
04.04.01 «Chemistry»
field of studies / speciality code and title
The course instruction is implemented within the professional education programm
of higher education:
«Fundamental and applied chemistry»
higher education programme profile/specialisation title

1. COURSE GOAL(s)

The goal of the course of «Chemistry of natural compounds» is in:

- familiarization with the properties, structure and synthesis of molecules of organic compounds found in wildlife;
 - study of the relationship of structural formulas with their biological functions;
- getting an idea about the most important classes of natural compounds that play a significant role in the life of plants and animals;
- disclosure of the fundamental mechanisms of action, biosynthesis and industrial synthesis of vitamins and hormones on the human body;
- study at the molecular level of the transmission of hereditary information (amino acids, proteins, nucleic acids);
- discussion of the most important processes occurring in plants: photosynthesis, the formation of terpenoids and alkaloids.

2. REQUIREMENTS FOR LEARNING OUTCOMES

Mastering the discipline "Chemistry of natural compounds" is aimed at developing the following competences (competences in part):

Table 2.1. List of competences that students acquire through the course study

Competence code	Competence descriptor	Competence formation indicators (within this course)
PC-1	The ability to plan work and choose adequate methods for solving research problems in the chosen field of chemistry, chemical technology or sciences related to chemistry	

3.COURSE IN HIGHER EDUCATION PROGRAMME STRUCTURE

The course refers to the elective component of (B1) block of the higher educational programme curriculum.

Within the higher education programme students also master other (modules) and / or internships that contribute to the achievement of the expected learning outcomes as results of the course study.

Table 3.1. The list of the higher education programme components/disciplines that contribute to the achievement of the expected learning outcomes as the course study results

Competenc	Competence	Previous	Subsequent
e code	descriptor	courses/modules*	courses/modules*
PC-1	The ability to plan	Methods of Organic	Undergraduate practice
	work and choose	Chemistry	
	adequate methods for	Fundamentals of	
	solving research	biotechnology	
	problems in the chosen	Molecular spectral	
	field of chemistry,	analysis Domino reactions	

Competenc	Competence	Previous	Subsequent
e code	descriptor	courses/modules*	courses/modules*
		in the synthesis of	
		heterocycles Research	
	chemical technology	work	
	or sciences related to	Experimental methods in	
	chemistry	the chemistry	
		NMR of organic	
		compounds	

^{*} To be filled in according to the competence matrix of the higher education programme.

4. COURSE WORKLOAD AND ACADEMIC ACTIVITIES

The total workload of the course is 3 credits

Table 4.1. Types of academic activities during the periods of higher education

programme mastering (full-time training)

Types of academic activities		Total as h	Semester(-s)			
		Total, ac.h.	1	2	3	4
Contact academic hours		36			36	
including:						
Lectures (LC)		36			36	
Lab works (LW)						
Seminars (workshops/tutorials) (S)						
Self-studies		54			54	
Evaluation and assessment (exam/passing/failing grade)		18			18	
Course workload	academic hours	108			108	
	credits	3			3	

5. COURSE CONTENTS

Table 5.1. Course contents and academic activities types

Course module title	Course module contents (topics)	Academic activities types
Section 1. Introduction.	Theme 1.1. Organic compounds of human and animal organisms - proteins, nucleic acids, lipids; plants - carbohydrates, alkaloids. Biosphere and its development. Theoretical and practical significance of studying the chemistry of natural compounds.	LC
Section 2. Proteins.	Theme 2.1. Protein content in the body and their functions. Variety of natural proteins. The beginning of protein chemistry. Qualitative reactions of proteins. The primary structure of proteins. Secondary structure of proteins. Tertiary structure of a proteins. Polypeptide synthesis.	LC
Section 3. Nucleic acids.	Theme 3.1. Deoxyribose nucleic acid is the human genome. The structure of the macromolecule, DNA, gene and other parts. DNA is the primary template for the synthesis of	LC

	proteins in each individual. Cell plasma is matrix ribose nucleic acid, mRNA. Protein synthesis. One gene, one protein. Discovery of DNA (I. Miter). The structure of a section of DNA - a gene. Nucleotides, nucleosides, D - 2 - deoxyribose furanose, D - ribose furanose, purine and pyrimidine bases. Primary structure of NC. Nucleotide sequence. A triplet of nucleotides is the code of a - amino acids. Secondary structure of DNA. Double helix of nucleotide chains with antiparallel orientation.	
Section 4. Enzymes (enzymes).	Theme 4.1. Enzymes are biocatalysts. Biochemical enzymatic processes - baking, alcoholic fermentation (yeast). Simple enzymes. Complex proteins. Anoenzyme and coenzyme. Enzymatic reactions. Nicotinamidin dinucleotide NAD.	LC
Section 5. Lipids (fats).	Theme 5.1. Lipids are the building material of cell membranes and various body tissues; a source of energy that ensures the vital activity, growth and development of the body. Neutral (simple) triglycerides. Isomerism of triglycerides according to the position of acyl radicals. Targeted synthesis of triglycerides.	LC
Section 6. Vitamines	Theme 6.1. The biological role of vitamins. Avitaminosis and hypervitaminosis. Classification of vitamins. fat soluble vitamins. Structure and chemical synthesis. Vitamins A, D, K, E. Water-soluble vitamins. Structure, biological role and synthesis. Vitamins C, group of vitamins B, vitamin PP. Mechanisms of the biological action of vitamins.	LC
Section 7. Terpenes and	Theme 7.1. Classification, finding in nature, methods of isolation from natural sources.	LC
terpenoids.	Synthesis and application.	
Section 8. Hormones.	Theme 8.1. Hormones, physiologically active substances - regulators of biochemical metabolic processes in the body. Hormones of a number of biogenic a-amino acids. Thyroxin. Synthesis of thyroxine. Thyroxine analogues: thyroidin, diiodothyrosine, betazine. Synthesis of these hormones. Hormones of a number of amino alcohols: Adrenaline, Serotonin, Histamine (synthesis, biosynthesis, functions). Nonapeptides, vasopressin and oxytocin. The difference is in the structure of the peptide chain and in the physiological action. Samotropin. Insulin (Structure. Biotechnological production of insulin). Steroid hormones. Pregnane derivative - cortisone. Industrial production of cortisone from sapogenin. Sex	LC

	hormones. Androstan derivatives, male sex	
	hormones: testosterone, androsterone,	
	dehydroandrosterone. Estrane derivatives, female	
	sex hormones: estrone, estradiol, estrol.	
	Cholesterol is the basis for the synthesis of steroid	
	hormones in the body. Industrial methods for	
	obtaining steroid hormones.	
	Theme 9.1. Alkaloids are organic compounds	LC
	found in various parts of a plant. Piperidine	
	(coniine, arecoline, lobeline and its analogues),	
	Pyridine (nicotine, anabasine), Tropanic (tropine,	
	atropine, scopolamine, cocaine, cynamylcocaine),	
	imidazole alkaloid pilocarpine, Quinoline	
	alkaloids. Industrial method for the synthesis of	
	papaverine. Purine alkaloids are tea alkaloids.	
	Industrial method for the synthesis of	
	theophylline Caffeine is a psychotropic agent, a	
	tonic. Quinolizidine alkaloids: lupinine,	
Section 9. Alkaloids.	pachycarpine, cytisine. Condensed systems with a	
	fragment of quinolizidine.	
	Pyrrolysine alkaloids, esters of the dihydric	
	alcohol of platinecin - platifillin and sarracin.	
	Polycyclic condensed alkaloids: berberine,	
	emytine, ezerine, reserpine, morphine. Morphine	
	derivatives: codeine and heroin. Morphine is an	
	effective pain reliever (analgesic). Morphine and	
	heroin are strong drugs. Acyclic alkaloids	
	(phenethylamine): tyramine, mexalin, ephedrine.	
	Industrial synthesis of ephedrine. The steroidal	
	alkaloids salosidin and salonidin.	
	Theme 10.1. The essence and significance of	LC
	photosynthesis for the development of the Earth.	
	History of the doctrine of photosynthesis. The	
Section 10.	scale of photosynthetic activity in the biosphere.	
Section 10.	Basic and additional photosynthetic pigments:	
Photosynthesis.	chlorophylls, carotenoids, phycobilins.	
	Chloroplasts and their role in photosynthesis. The	
	structure of chloroplasts. Chlorophyll a and	
	chlorophyll b. The porphine structure is the basis	
	of the molecular structure of the chloroplast.	

^{* -} to be filled in only for <u>full</u> -time training: LC - lectures; LW - lab work; S - seminars.

6. CLASSROOM EQUIPMENT AND TECHNOLOGY SUPPORT REQUIREMENTS

Table 6.1. Classroom equipment and technology support requirements

Type of academic activities	Classroom equipment	Specialised educational / laboratory equipment, software, and materials for course study (if necessary)
Lecture	A lecture hall for lecture-type classes,	

Type of academic activities	Classroom equipment	Specialised educational / laboratory equipment, software, and materials for course study (if necessary)
	equipped with a set of specialised furniture; board (screen) and technical means of multimedia presentations.	
Self-studies	A classroom for independent work of students (can be used for seminars and consultations), equipped with a set of specialised furniture and computers with access to the electronic information and educational environment.	

^{*} The premises for students' self-studies are subject to **MANDATORY** mention

7. RESOURCES RECOMMENDED FOR COURSE STUDY

Main readings:

- 1. BIOCHEMISTRY 2nd ed., corrected. and additional Textbook and workshop for open source software. Scientific school: Moscow State Technical University named after N.E. Bauman (Moscow). Ershov Yu. A., Zaitseva N. I.; Ed. Schukina S.I. https://www.biblio-online.ru/book/biohimiya-442538
- 2. BIOCHEMISTRY FOR TECHNOLOGIES IN 2 H. PART 2 2nd ed. Textbook and workshop for secondary vocational education 1 Scientific school: Vologda State Dairy Academy named after NV Vereshchagin (Vologda). https://www.biblio-online.ru/book/biohimiya-dlya-tehnologov-v-2-ch-chast-2-442534
- 3. BIOLOGICAL CHEMISTRY 2nd ed., Per. and additional Textbook for bachelor's, specialist's and master's programs 2. Scientific school: Crimean Federal University named after V.I. Vernadsky (Simferopol). Dryuk V. G., Sklyar S. I., Kartsev V. G. https://www.biblio-online.ru/book/biologicheskaya-himiya-442129
 https://www.biblio-online.ru/book/biologicheskaya-himiya-442129
 https://www.biblio-online.ru/book/biologicheskaya-himiya-442129
- 1. Chemistry of natural compounds (carbohydrates, nucleides, steroids, proteins). Kochetkov N.K., Torgov I.V., Botvinik M.M. 1961. (http://biblioclub.ru/index.php? page=book view red&book id=430998)
- 2. Fundamentals of the chemistry of natural compounds (volumes 1 and 2). Semyonov A.A., Kartsev V.G., Moscow, 2009
- 3. Soldatenkov A.T., Kolyadina N.M., Shendrik. I.V. Fundamentals of organic chemistry of medicinal substances Chemistry, Moscow, 2003.
- 4. V.V. Plemenkov, Introduction to the chemistry of natural compounds. Kazan, 2001
- 5. Chemistry of biologically active compounds (Theory and Practice): textbook 2018 Bolotov V. M. Komarova E. V. Savvin P. N. Publisher: Voronezh State University of Engineering Technologies 85 pp. ISBN: 978-5-00032-306- 9 BBK: G2 i7 UDC: 577.15/.17 (http://biblioclub.ru/index.php?page=book red&id=487998)

Internet sources

- 1. Electronic libraries (EL) of RUDN University and other institutions, to which university students have access on the basis of concluded agreements:
 - RUDN Electronic Library System (RUDN ELS) http://lib.rudn.ru/MegaPro/Web
 - EL "University Library Online" http://www.biblioclub.ru
 - EL "Yurayt" http://www.biblio-online.ru

- EL "Student Consultant" www.studentlibrary.ru
- EL "Lan" http://e.lanbook.com/
- EL "Trinity Bridge"

DEVELOPERS:

Position, Department

- 2. Databases and search engines:
- Yandex search engine https://www.yandex.ru/
- Google search engine https://www.google.ru/
- abstract database SCOPUS http://www.elsevierscience.ru/products/scopus/ Training toolkit for self- studies to master the course *:
- 1. Course of lectures on the discipline "Chemistry of Natural Compounds".
- 2. Guidelines for the study of the special course "Chemistry of natural compounds" Chapters 1-4. Lipids, hormones, plant protection chemicals. RUDN University, Moscow, 1987.
- 3. Guidelines for the study of the special course "Chemistry of natural compounds" Chapters 5-8. Protein substances, nucleic acids, photosynthesis, organic drugs. RUDN University, Moscow, 1987.
- 4. Varlamov A.V., Borisova T.N., Sorokina E.A., Voskresensky L.G., Nikitina E.V. Fundamentals of organic chemistry. M .: RUDN University, 2007. 356.
- * The training toolkit for self- studies to master the course is placed on the course page in the university telecommunication training and information system under the set procedure.

Associate Professor of the Department of Organic E.V. Nikitina Chemistry Position, Department Signature name and surname **HEAD OF EDUCATIONAL DEPARTMENT:** L. G. Voskressensky **Organic Chemistry Department** Name of Department ignature name and surname **HEAD OF HIGHER EDUCATION PROGRAMME:** Dean of Science faculty, Head of the Department of L. G. Voskressensky **Organic Chemistry**

Signature

name and surname